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Rate of net energy transfer by heat, work, and mass. system (steady) Rate of change in internal, kinetic, potential, etc. energies. out (since ke pe 0) 12434 124443444 Δ. ΔΔ. Ê 0. 12 12. 0. Substituting, Q&out= &Cm pΔT=() 120 kg/min(1.007kJ/kg °C)(°C3)= 363 kJ/min. 120 kg/min AIR. Q · Chapter 1 Basics of Heat Transfer

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1-2. Thermodynamics and Heat Transfer. 1-1C Thermodynamics deals with the amount of heat transfer as a system undergoes a process from one equilibrium state to another.

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Assumptions Heat transfer from the surface of the filament and the bulb of the lamp is uniform . Analysis (a) The heat transfer surface area and the heat flux on the surface of the filament are $A_{DL} = \pi \pi = (0.05 \text{ cm})(5 \text{ cm}) 0.785 \text{ cm}^2$ $s = = 2 = \times 1.91 \text{ } 10 \text{ W/m}^2 2 \text{ } 191 \text{ W/cm}^2 0.785 \text{ cm}^2 150 \text{ W}$ $s s A Q q \& \& (b) \text{ The heat flux on the surface of glass bulb is}$

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1-13C The rate of heat transfer per unit surface area is called heat flux q . It is related to the rate of heat transfer by $^3 A Q qdA$. 1-14C Energy can be transferred by heat, work, and mass. An energy transfer is heat transfer when its driving force is temperature difference.

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... least in the aspects of heat and/ or mass transfer, analytical/numerical solutions and optimization of the processes The first chapter deals with the description and mass transfer analysis of fixed-bed ... layer on the heat transfer process in high temperature polymerization Chapter is concerned with the description and analysis of heat and mass transfer processes in a fin -and- tube air heater Hydrogen ...

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