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Tuesday 8 May 2012 (afternoon)

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INSTRUCTIONS TO CANDIDATES • Do not open this examination paper until instructed to do so. • Answer all the questions. • For each question, choose the answer you consider to be the best and indicate your choice on the answer sheet provided.

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points than the total allows This is
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1.01 §. ① Do not penalize if integer values of atomic masses used. Accept alternative calculation method. Award [2] for three correct. Award [1] for any two correct. PbC 8 H 20; [3] (ii) PbC 8 H 20; [1] (iii) PbC H 14O € €PbO 8CO 10H O 8 20 2 2 2 2 o correct reactants and products; [2] correct coefficients; M2 can only be scored ...

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CIE Paper 1 IGCSE Chemistry Past Papers - PMT

Paper 3 (SL): 1 hour. This paper will have questions on core and SL option material. Section A: one data-based question and several short-answer questions on experimental work. Section B: short-answer and extended-response questions from one option. The

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questions on paper 3 test assessment objectives 1, 2 and 3.

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M12/4/CHEMI/HP3/ENG/TZ1/XX/M 20
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37 Rb 8547 55 Cs 13291 87 Fr (223)
Atomic number Element Relative atomic
mass 25 Mn 5494 43 Tc 9891 75 Re
18621 61 Pm 14692 93 Np (237)

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